

The comparative tissues distribution of platinum and ^{14}C in mice receiving ^{14}C -labelled carboplatin*

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Summary. Since the reactivity of carboplatin depends on the rate of removal of the 1,1-cyclobutanedicarboxylate ligand, the time course of this dissociation has been determined in various tissue and body fluids using ^{14}C -labelled carboplatin (*cis*-diammine[1,1-cyclobutane-1- ^{14}C -dicarboxylate]platinum II). Mice received ^{14}C -carboplatin (80 mg/kg; 1.1 mCi/kg, i.v.), and tissue was removed at times ranging from 5 min to 5 days posttreatment. Following solubilization, tissue aliquots were analyzed for platinum and ^{14}C contents. Carboplatin remained intact for up to 2 h posttreatment, since the ratio of ^{14}C :Pt in tissues (nmol/g) was unity. Thereafter, the ^{14}C ligand was released from the molecule and preferentially removed from tissues, indicated by decreasing ^{14}C :Pt ratios. The elimination half-lives for Pt varied between tissues (40–156 h). In contrast, the corresponding half-lives for the ^{14}C species were similar in most types of tissue (18–35 h), although those in the liver and spleen were exceptional (210 and 90 h, respectively). At 5 days a maximum of 4%–24% of the total Pt in tissue might exist as intact drug. Thus, the metabolic handling of carboplatin varies according to the tissue, since the elimination of the ^{14}C cyclobutane dicarboxylate species from most tissue was similar and Pt elimination was slower and tissue-dependent.

Introduction

Cisplatin (*cis*-dichlorodiammine platinum II) is a well-established agent in cancer chemotherapy, although its side effects, notably dose-limiting nephrotoxicity, have hindered its full clinical exploitation [9, 11, 17]. Carboplatin was developed as a more stable, non-nephrotoxic analogue of cisplatin [5]. Phase I evaluation revealed myelosuppression to be dose-limiting and confirmed an absence of nephrotoxicity at the recommended phase II dose [1]. Subsequent clinical studies in other centers have confirmed these findings [4]. Phase II studies have revealed activity in ovarian [23], small cell lung [22], and testicular [16]

tumors, and the drug was registered in the United Kingdom for the two former indications in 1986. The differential nephrotoxicities of cisplatin and carboplatin have been attributed to differences in the mechanisms of their renal handling [20]. The chemical reactivity of the complex may also be an important determinant of nephrotoxicity [3]. However, both cisplatin and carboplatin are inert as intact molecules, and their activation must occur before any biological activity is manifested. The activation of cisplatin is chemically mediated and requires aquation of the molecule [10], the first aquation step probably being rate-limiting in the platination of macromolecules [3, 10]. Carboplatin is thought to be activated in a similar manner, although the presence of the bidentate cyclobutane dicarboxylate ligand, in place of the two chloride ligands of cisplatin, renders the former compound 10- to 20-fold less reactive [8, 20]. This, however, does not discount the possibility of enzyme-mediated activation of carboplatin, as proposed by Cleare [2]. The rate of activation in a tissue may be related to its chemical and biochemical composition, since we have previously demonstrated that the binding of both carboplatin and cisplatin in the liver and kidney in vitro is significantly greater than that in the plasma [20].

In the present investigation, we have examined the abilities of tissue to activate carboplatin in vivo. This was achieved through synthesis of ^{14}C -carboplatin, where the label is present in the cyclobutane dicarboxylate ligand (Fig. 1), and by following separately the pharmacokinetics of Pt and ^{14}C species.

Materials and methods

Chemicals

Carboplatin (*cis*-diamminecyclobutanedicarboxylate platinum II, JM8, Paraplatin) a gift from the Johnson Matthey Research Centre (Reading, Berks, UK) was dissolved in 5% dextrose. ^{14}C -cyclobutane dicarboxylic acid was obtained from Amersham UK (Amersham, Bucks, UK). Hyamine hydroxide (methylbenzethonium hydroxide in 1 M methanol) was obtained from Sigma Chemicals (UK).

Methods

^{14}C -carboplatin synthesis. 1,1-cyclobutane- ^{14}C -dicarboxylic acid was synthesized by Amersham (UK) and had a specific activity of 68 $\mu\text{Ci}/\text{mg}$. ^{14}C -carboplatin was synthesized

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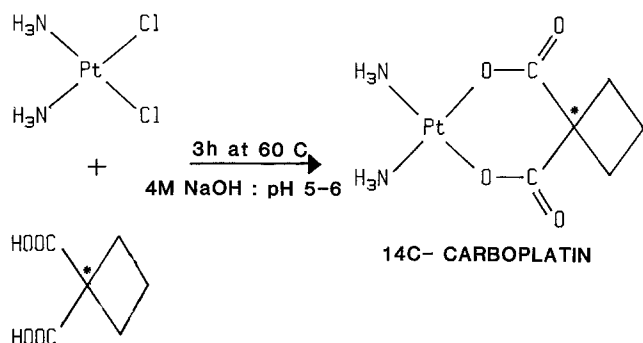


Fig. 1. Synthesis of *cis*-diammine-1,1-cyclobutane-1-¹⁴C-dicarboxylate platinum II. * indicates the position of the ¹⁴C atom

as described by the Research Corporation [7], as shown in Fig. 1. Briefly, silver nitrate (3.2 mmol) was dissolved in 1 ml distilled water. Platinum diiododiammine (1.66 mmol) was added with rapid stirring and maintained at 40° C for 1 h in a dark environment. The silver iodide precipitate was filtered off, and activated charcoal was added to the filtrate with 4 × 0.5 ml rinses of distilled water. ¹⁴C-cyclobutane dicarboxylic acid (2.11 mmol) was dissolved in 0.75 ml 4 M sodium hydroxide, the final solution having a pH of 5–6. The diaquo Pt filtrate was added to ¹⁴C-cyclobutane dicarboxylic acid with 0.5 ml water and the reaction maintained at 60° C for 3 h with formation of a white precipitate. The temperature was closely monitored to prevent decomposition of the product (> 60° C). The white precipitate was isolated by filtration, rinsed with 1 ml water and ethanol, and then reheated to 60° C in a flask with 1 ml distilled water. The solution was refiltered into a flask for freeze-drying over 24 h. The crude yield (52% yield) of ¹⁴C-carboplatin was 0.87 mmol.

Analysis of ¹⁴C-carboplatin purity. The chemical purity of the compound (> 96%) was assessed using flameless atomic absorption spectrophotometry. Its radiochemical purity was assessed by spotting 0.2 nmol ¹⁴C-carboplatin (in water) onto silica gel-treated (60 F₂₅₄) thin-layer chromatography plates (Merck, Germany) and chromatographing for 30 min in acetone:water (4:1). After drying, the plates were scanned at 254 nm using a computerized Bertrol system, and 98.56% radioactivity (¹⁴C-carboplatin) was located in a single peak. The specific activity of the product was assessed by liquid scintillation counting, using 8 ml cocktail T scintillant (BDH, UK) and ¹⁴C-hexadecane as the external standard (Amersham, UK; specific activity 0.39 μCi/ml). The counting efficiency was > 96%. The specific activity of ¹⁴C-carboplatin was 24.07 μCi/mg (4.66 μCi/mmol).

Tissue distribution studies. Female Balb/C— mice (18–20 g) received 20–25 μCi of ¹⁴C-carboplatin (80 mg/kg; 1.1 mCi/kg; 10 ml/kg) or ¹⁴C-cyclobutane dicarboxylic acid (31 mg/kg; 1.1 mCi/kg; 10 ml/kg). The dose solutions comprised radiolabelled drug diluted to the required dose with “cold” unlabelled drug in order to obtain the specific activities quoted.

Mice were anesthetized with diethylether and exsanguinated at 5 min, 1, 4, and 8 h, 1, 2, and 5 days post treatment. Tissue (liver, kidney, muscle, ileum, heart, lung, uterus, skin, spleen, plasma, blood) was removed, rinsed

in saline, gently blotted dry, and weighed. Approximately 200 mg tissue was solubilized overnight at 60° C in 500 μl hyamine hydroxide and diluted in 0.1 N HCl to a total volume of 5.5 ml. Aliquots of this were analyzed for ¹⁴C by standard liquid scintillation techniques. Cocktail T scintillant (8 ml) and 25 μl glacial acetic acid (to reduce the chemiluminescence of hyamine) were added to 0.2- to 1-ml hyamine-solubilized tissue digests. Aliquots (0.1–1.5 ml) of hyamine-solubilized liver (spiked with ¹⁴C standard) were used to determine a colour quench correction curve. Samples were counted by the method of external standardization; the counting efficiency was > 82%. Platinum analysis was carried out by flameless atomic absorption spectrophotometry as previously described by Siddik et al. [21].

Platinum and ¹⁴C concentrations are expressed as nmol/g in tissue or nmol/ml in fluid. These respective concentrations were used to calculate the ¹⁴C:Pt ratios in the tissues and fluids. Thus, a ratio of 1 in tissue indicated that carboplatin was probably in its intact form, as in the dose solution. Similarly, a ratio of < 1 would indicate a greater loss of ¹⁴C than of Pt from the tissue.

Metabolic studies. Mice injected i.v. with ¹⁴C-carboplatin (80 mg/kg; 1.1 mCi/kg) or ¹⁴C-cyclobutane dicarboxylic acid (31 mg/kg; 1.1 mCi/kg) were housed in glass Metabowls (Jencons Scientific, Leighton Buzzard, UK) for 48 h. Exhaled CO₂ was trapped in 1-ethanolamine:2-methoxyethanol (1:2) and samples were analyzed for ¹⁴C at 6 h and at 1 and 2 days posttreatment as described by Siddik et al. [19]. Flameless atomic absorption spectrophotometry was used to determine Pt levels in urine and in faeces digested in concentrated nitric acid. For faecal ¹⁴C analysis, samples were oxidized using a Packard oxidizer (recovery > 98%) prior to counting by liquid scintillation.

Pharmacokinetic analysis. The plasma or tissue concentration-time data for both Pt and ¹⁴C were analyzed by a two-compartment, open model using a nonlinear least squares computer program [18], as described by the equation:

$$C_0 = A_{\text{exp}} e^{(-\alpha t)} + B_{\text{exp}} e^{(-\beta t)}.$$

Results

All tissues of mice receiving ¹⁴C-carboplatin (80 mg/kg; 1.1 mCi/kg) showed a biphasic decline in both ¹⁴C and Pt levels over 5 days, as exemplified by the plasma, liver, and kidney in Fig. 2. At 5 min post treatment, Pt concentrations in the plasma, liver, and kidney were 446, 138, and 1405 nmol Pt/ml or per gram (Table 1). These concentrations fell rapidly with time. The initial rate of loss of Pt and ¹⁴C from all tissues (over the 2 h) was similar. Thereafter, the rate of removal of ¹⁴C exceeded the rate of Pt removal (Table 2). In the liver, the pattern of ¹⁴C removal differed from that in all other tissue, since the ¹⁴C curve levelled out in the terminal phase.

The distribution of ¹⁴C-cyclobutane dicarboxylic acid (31 mg/kg; 1.1 mCi/kg) was studied to compare the tissue distribution and elimination phase half-lives with those of ¹⁴C-carboplatin. The highest ¹⁴C-cyclobutane dicarboxylic acid levels were observed in the kidney (Table 3) and may be due to the high urinary excretion of this species. The half-lives in plasma and tissue for the two compounds differed (Table 4). Interestingly, there was no substantial dif-

Table 1. Platinum concentrations (nmol/g) in tissues of mice receiving ^{14}C -Carboplatin (80 mg/kg; 1.1 mCi/kg)

Tissue	5 min	1 h	4 h	8 h	1 day	2 days	5 days
Liver	138 \pm 26.1	108 \pm 20.5	34.9 \pm 2.0	28.8 \pm 3.1	26.7 \pm 0.3	19.8 \pm 2.6	13.2 \pm 1.6
Kidney	1405 \pm 107	158 \pm 92.3	46.1 \pm 4.6	30.8 \pm 7.8	37.4 \pm 18.5	26.8 \pm 6.15	12.3 \pm 2.6
Muscle	77 \pm 8.3	17.1 \pm 4.1	6.6 \pm 0.4	6.9 \pm 0.6	2.6 \pm 0.1	3.2 \pm 0.3	1.7 \pm 0.1
Ileum	66 \pm 5.1	45.7 \pm 6.1	19.8 \pm 3.1	15.7 \pm 2.6	7.2 \pm 1.1	4.5 \pm 1.0	2.3 \pm 0.5
Heart	117 \pm 16.1	16.5 \pm 1.9	6.5 \pm 0.6	4.4 \pm 0.4	3.2 \pm 0.2	2.9 \pm 0.5	2.3 \pm 0.6
Lung	241 \pm 17.3	35.9 \pm 1.9	16.0 \pm 1.6	13.7 \pm 1.4	6.6 \pm 0.3	5.8 \pm 0.8	4.1 \pm 0.3
Uterus	287 \pm 23.1	33.6 \pm 2.6	16.6 \pm 2.2	9.4 \pm 1.3	7.3 \pm 0.6	5.1 \pm 0.1	12.9 \pm 2.4
Skin	246 \pm 42.0	36.9 \pm 3.3	16.5 \pm 2.1	16.4 \pm 1.5	8.9 \pm 1.1	10.4 \pm 0.5	9.3 \pm 0.3
Spleen	77 \pm 7.8	38.3 \pm 5.5	15.0 \pm 1.8	18.0 \pm 2.0	16.8 \pm 2.0	18.4 \pm 1.0	18.7 \pm 2.5
Plasma	446 \pm 201	45.7 \pm 4.6	4.9 \pm 0.3	3.9 \pm 0.1	2.5 \pm 0.2	1.7 \pm 0.1	0.6 \pm 0.05
Blood	195 \pm 88.1	83.0 \pm 11.3	24.5 \pm 4.2	16.0 \pm 1.8	6.8 \pm 0.6	6.5 \pm 0.6	6.9 \pm 0.51

Results are expressed as $x \pm \text{SEM}$, $N = 4$

Table 2. ^{14}C concentrations (nmol/g) in tissues of mice receiving ^{14}C -carboplatin (80 mg/kg; 1.1 mCi/kg)

Tissue	5 min	1 h	4 h	8 h	1 day	2 days	5 days
Liver	150 \pm 13.3	102 \pm 50.3	18.8 \pm 0.9	10.6 \pm 0.5	7.5 \pm 0.5	5.9 \pm 0.2	5.8 \pm 0.2
Kidney	1555 \pm 112	171 \pm 37	26.3 \pm 1.7	14.3 \pm 0.6	5.8 \pm 0.9	1.8 \pm 0.1	0.5 \pm 0.01
Muscle	59 \pm 6.3	8.2 \pm 0.4	3.4 \pm 0.2	2.9 \pm 0.2	0.9 \pm 0.2	0.5 \pm 0.05	0.2 \pm 0.03
Ileum	136 \pm 7.8	48.0 \pm 2.8	11.9 \pm 0.3	6.5 \pm 0.4	1.8 \pm 0.4	0.6 \pm 0.07	0.2 \pm 0.07
Heart	91.4 \pm 10.1	14.3 \pm 1.5	4.3 \pm 0.1	2.7 \pm 0.1	0.9 \pm 0.07	0.5 \pm 0.05	0.2 \pm 0.03
Lung	244 \pm 21.3	39.8 \pm 3.6	13.9 \pm 1.2	9.2 \pm 0.5	2.3 \pm 0.2	0.9 \pm 0.07	0.2 \pm 0.02
Uterus	289 \pm 13.8	40.5 \pm 5.9	15.0 \pm 0.9	6.9 \pm 0.4	1.8 \pm 0.4	0.9 \pm 0.1	0.3 \pm 0.02
Skin	194 \pm 40.9	26.9 \pm 1.8	10.4 \pm 0.1	5.8 \pm 0.3	1.9 \pm 0.1	1.0 \pm 0.05	0.3 \pm 0.01
Spleen	71 \pm 8.6	33.8 \pm 0.2	13.1 \pm 0.9	10.2 \pm 0.9	6.3 \pm 1.0	5.2 \pm 0.5	3.3 \pm 0.1
Plasma*	691 \pm 181	46.5 \pm 7.0	4.1 \pm 0.07	2.7 \pm 0.5	1.1 \pm 0.2	0.3 \pm 0.01	<0.1
Blood*	412 \pm 93.1	52.3 \pm 1.6	6.8 \pm 0.3	4.5 \pm 0.08	1.8 \pm 0.08	0.6 \pm 0.01	<0.1

Results are expressed as $x \pm \text{SEM}$, $N = 4$

* Plasma and blood ^{14}C concentrations at 5 days were <0.1 nmol/ml and have been excluded from calculations of terminal half-lives

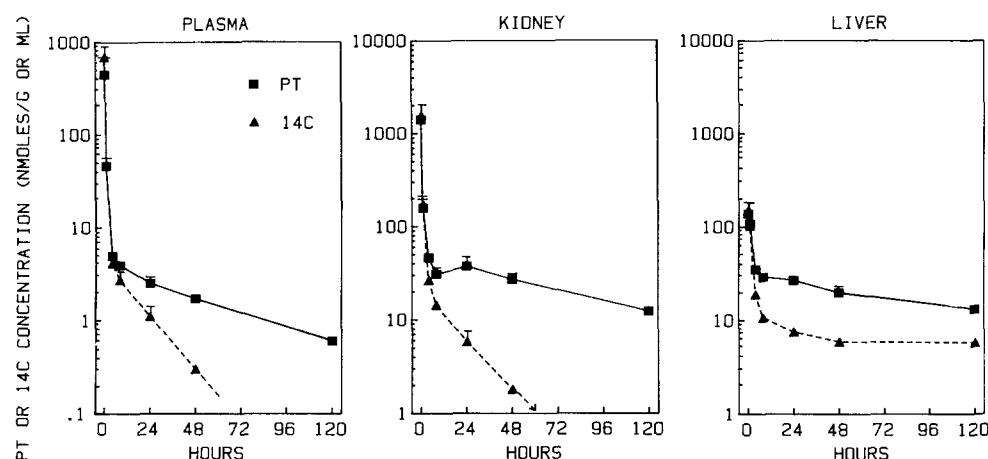


Fig. 2. Decay of Pt (■) and ^{14}C (▲) in plasma, kidney, and liver following ^{14}C -carboplatin (80 mg/kg; 1.1 mCi/kg; i.v.) over 5 days. Results are expressed as mean \pm SEM, $N = 4$

ference in the retention or elimination half-lives of ^{14}C -cyclobutane dicarboxylic acid in liver compared with that in other tissues (Tables 3 and 4).

The elimination phase half-lives for Pt varied between different types of tissue, from 40 h in plasma to 156 h in skin (Table 4). The corresponding half-lives for ^{14}C were similar for most tissue (18–35 h), although the liver and spleen were exceptions (210 and 90 h, respectively).

The temporal changes in the normalized ^{14}C :Pt ratios are illustrated for four types of tissue in Fig. 3. A ratio of unity was demonstrated at 1 h and maintained for up to 2 h in all tissues. In most types of tissue the ratio decayed rapidly thereafter, whereas in the liver the ratio at 5 days was 2- to 10-fold greater than in other tissues. At 5 days, the normalized ^{14}C :Pt ratios in most tissues were 0.04–0.24. In the liver, on the other hand, only 48% of the

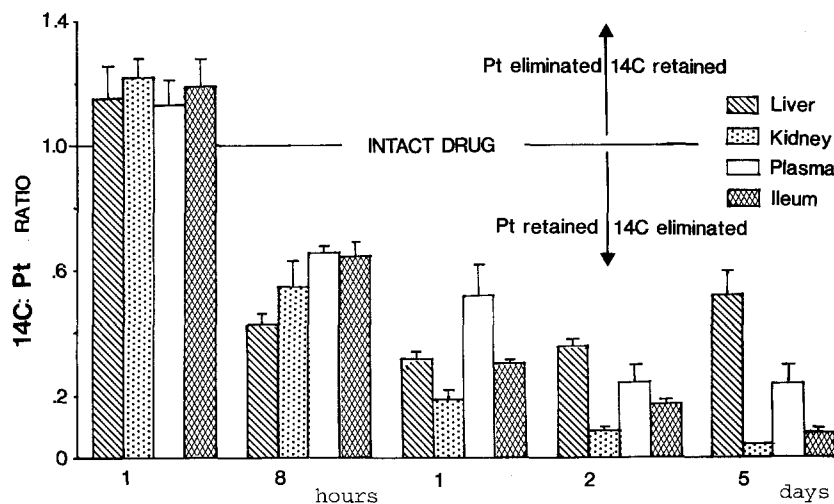


Fig. 3. Tissue ^{14}C :Pt ratios in four tissues. Intact carboplatin is represented by a ratio of unity. Results are expressed as mean \pm SEM, $N = 4$

Table 3. ^{14}C concentrations (nmol/g) in tissues of mice receiving 1,1- ^{14}C -cyclobutane dicarboxylic acid

Tissue	5 min	1 h	4 h	8 h	1 day	2 days	5 days
Liver	1078 \pm 333	29.1 \pm 8.9	3.7 \pm 0.4	2.0 \pm 0.2	1.7 \pm 0.06	1.4 \pm 0.02	1.0 \pm 0.01
Kidney	1235 \pm 197	380 \pm 166	14.9 \pm 4.5	7.2 \pm 0.3	5.5 \pm 0.6	4.8 \pm 0.4	2.4 \pm 0.07
Muscle	66.9 \pm 4.9	13.8 \pm 3.9	1.2 \pm 0.3	0.2 \pm 0.01	0.2 \pm 0.02	0.1 \pm 0.01	0.1 \pm 0.01
Ileum	58.5 \pm 12.5	21.0 \pm 5.9	3.3 \pm 0.6	0.9 \pm 0.07	0.6 \pm 0.03	0.5 \pm 0.02	0.2 \pm 0.01
Heart	110 \pm 3.8	25.1 \pm 7.26	0.9 \pm 0.3	0.3 \pm 0.01	0.3 \pm 0.03	0.3 \pm 0.03	0.2 \pm 0.01
Lung	216 \pm 10	56.7 \pm 20	2.9 \pm 3.1	0.7 \pm 0.07	0.3 \pm 0.02	0.3 \pm 0.01	0.1 \pm 0.01
Uterus	140 \pm 29	44.3 \pm 11.4	10.7 \pm 3.1	0.9 \pm 0.07	1.2 \pm 0.1	0.5 \pm 0.09	0.2 \pm 0.05
Skin	144 \pm 4.9	25.4 \pm 8.9	0.9 \pm 0.4	0.4 \pm 0.02	0.4 \pm 0.04	0.8 \pm 0.2	0.3 \pm 0.01
Spleen	71 \pm 2.4	25.5 \pm 7.3	4.2 \pm 0.9	1.4 \pm 0.2	0.6 \pm 0.04	0.6 \pm 0.03	0.3 \pm 0.01
Plasma*	412 \pm 129	241 \pm 146	2.2 \pm 0.8	0.3 \pm 0.01	0.1 \pm 0.01	0.1 \pm 0.01	<0.1
Blood*	388 \pm 14.5	64.8 \pm 21.0	1.9 \pm 0.9	0.7 \pm 0.2	0.3 \pm 0.1	0.3 \pm 0.1	<0.1

Results are expressed as $x \pm \text{SEM}$, $N = 4$

* Plasma and blood ^{14}C concentrations at 5 days are <0.1 nmol/ml and have been excluded from calculations of terminal half-lives

Table 4. Elimination half-lives of platinum and ^{14}C in tissues of mice receiving ^{14}C -carboplatin or ^{14}C -cyclobutane dicarboxylic acid

Tissue	Half-life (h)		
	Carboplatin		^{14}C -CBDCA
	Pt	^{14}C	
Plasma	39.8 \pm 6.0	27.2 \pm 8.6	54.7 \pm 25.8
Liver	98.9 \pm 39.0	210.2 \pm 127.8	105.3 \pm 46.6
Kidney	70.7 \pm 23.6	25.8 \pm 6.4	74.1 \pm 26.7
Ileum	54.1 \pm 19.4	25.9 \pm 7.6	71.1 \pm 26.2
Muscle	116.4 \pm 83.0	31.5 \pm 10.2	150.8 \pm 72.3
Heart	43.5 \pm 10.8	27.1 \pm 8.2	97.2 \pm 73.3
Spleen ^a	—	90.3 \pm 30.0 ^b	78.4 \pm 35.8
Skin ^a	156.4 \pm 100.6	28.8 \pm 4.4	—
Blood	145.8 \pm 98.4	17.6 \pm 1.6	27.5 \pm 16.8
Uterus ^a	—	34.8 \pm 7.0	83.7 \pm 30.8
Lung	143.2 \pm 100.0	31.5 \pm 5.6	70.2 \pm 33.0

Results are expressed as $x \pm \text{SD}$; $N = 4$

^a Half-lives could not be determined due to absence of a well-defined decay in these tissues over 2–120 h

^b ^{14}C Half-lives significantly ($P \leq 0.01$) longer than in other tissues

carboplatin could be ascribed to transformed Pt species. The extent of transformation in this tissue on day 1, however, was about 65%–70%, which was similar to that in the kidney and other tissues.

Metabolic studies

Excretion of ^{14}C -carboplatin (80 mg/kg; 1.1 mCi/kg) or ^{14}C -cyclobutane dicarboxylic acid (31 mg/kg; 1.1 mCi/kg) was primarily renal, with 86%–92% of the carboplatin and 96% of the ^{14}C -cyclobutane dicarboxylic acid eliminated in the urine at 4 days (Table 5). Faecal excretion accounted for 1% and 4% of ^{14}C and Pt, respectively, of the administered dose of ^{14}C -carboplatin. Low faecal excretion of ^{14}C -cyclobutane dicarboxylic acid was also obtained (3.5% dose). A negligible amount of the administered dose of ^{14}C -carboplatin (0.06%) and of ^{14}C -cyclobutane dicarboxylic acid (0.6%) was exhaled as $^{14}\text{CO}_2$.

Discussion

The chemical activation of cisplatin and possibly of carboplatin, through an aquation step, is thought to be rate-limiting in the subsequent platination process [3, 10]. The

Table 5. Cumulative excretion of carboplatin (80 mg/kg; 1.1 mCi/kg) and cyclobutane dicarboxylate ligand (31 mg/kg; 1.1 mCi/kg) over 4 days in mice

Time (h)	Urinary excretion (% dose)			CO ₂ excretion (% dose)	
	Carboplatin		¹⁴ C-CBDCA	Carboplatin	¹⁴ C-CBDCA
	Pt	¹⁴ C	¹⁴ C		
24	83.4 ± 2.7	88.1 ± 3.1	91.3 ± 4.9	0.049 ± 0.001	0.57 ± 0.03
48	85.1 ± 2.8	91.5 ± 4.0	93.8 ± 5.4	0.054 ± 0.001	0.60 ± 0.03
72	85.5 ± 2.8	91.8 ± 3.9	94.8 ± 4.8	0.057 ± 0.001	0.63 ± 0.03
96	85.6 ± 2.8	91.9 ± 4.1	95.7 ± 4.0	0.059 ± 0.01	0.65 ± 0.02

Results are expressed as $\bar{x} \pm \text{SE}$, $N = 4$

bidentate cyclobutane dicarboxylate ligand of carboplatin confers stability to the molecule and thereby renders carboplatin 10- to 20-fold less reactive than the parent, cisplatin [8, 20]. The rate of activation within a tissue may relate to its biochemical and chemical composition. In a previous study [20], we have demonstrated that binding of cisplatin and carboplatin to the liver and kidney significantly exceeded the rate of binding to plasma. Hence, through the synthesis of ¹⁴C-carboplatin it has been possible to examine the activation of carboplatin in tissue and to trace the fate of the Pt and ¹⁴C moieties.

The data in this communication indicate that carboplatin remains intact in plasma and tissue for up to 2 h. Thereafter the ligand is released from the molecule and preferentially removed from the tissue, as indicated by decreasing values for the ¹⁴C:Pt ratios in tissue. The decay of both Pt and ¹⁴C in all tissues is biphasic, with a rapid initial phase and a slow elimination phase. This is similar to the distribution pattern of Pt previously observed with cisplatin in rats and mice [6, 19]. Prolonged retention of platinum in tissue suggests irreversible or tight binding to tissue components. At 5 days posttreatment, Pt concentrations were greatest in the spleen, kidney, liver, and whole blood. High values for Pt in these organs following treatment with cisplatin have also been observed in rats [6, 12] and man [15]. Splenic Pt levels were 1.5-fold greater than kidney levels at 5 days, due to a continual uptake of Pt by this organ. The degradation of accumulated cell products such as Pt-containing erythrocytes may partly account for this phenomenon [13, 14].

The half-lives of Pt in tissue were highly variable (40–156 h), and may be due to varying rates of intrinsic repair processes and subsequent excision of the Pt lesion, together with varying rates of protein turnover at platination sites within tissue.

The activation of carboplatin would involve a Pt-O bond in nucleophilic displacement, resulting in the formation of a monodentate Pt species. Subsequent loss of the ¹⁴C ligand occurs, with a half-life of around 27 h. Since the ¹⁴C elimination half-lives in tissue were similar, the activation of carboplatin occurs at a similar rate in most of the tissue examined. Knox et al. [8] have shown that, following the initial binding of ¹⁴C-carboplatin to DNA, the ¹⁴C-cyclobutane dicarboxylic acid ligand is released from the isolated DNA with a half-life of 14.5 h, with no corresponding loss of Pt. Thus, the loss of ¹⁴C ligand occurs as a result of the fragmentation of the molecule and not through the loss of the bound drug. Of the remaining Pt in tissue, a

maximum of 4%–24% exists as intact drug and the rest is probably irreversibly bound. This assumes that all the ¹⁴C present in tissue at 5 days is intact carboplatin.

High hepatic ¹⁴C levels may be due to monodentate binding or the incorporation of ¹⁴C into endogenous compounds. In contrast to the liver, the kidney is an efficient tissue for ¹⁴C ligand elimination. However, "activation" of the molecule was similar in all tissue examined, suggesting that an enzymatic activation may not be involved. There was minimal transformation of ¹⁴C-carboplatin and ¹⁴C-cyclobutane dicarboxylic acid to ¹⁴CO₂.

The parallel study using ¹⁴C-cyclobutane dicarboxylic acid distribution did not clarify the fate of the ligand in carboplatin, since the ¹⁴C in the two compounds was handled differently in vivo. The major route of excretion of carboplatin was via the kidney, with a small percentage of the dose excreted in faeces.

These results imply that the metabolic handling of carboplatin varies according to the tissue. The elimination of the ¹⁴C-cyclobutane dicarboxylate species from tissues was similar. However, Pt elimination was generally slower and tissue-dependent.

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